Some features of RuCl₃-catalyzed periodate oxidation of 3-N-substituted 5-allenyl-2,5-dichloro-4,4-dimethoxycyclopent-2-en-1-ones

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3-N-Substituted 5-(1Z-carboxymethylene)-2-chloro-4,4-dimethoxycyclopent-2-en-1-ones have been prepared from the corresponding 3-N-substituted 5-allenyl-2,5-dichloro-4,4-dimethoxycyclopent-2-en-1-ones using selective oxidative cleavage of their allene bond by the RuCl₃-NaIO₄ system.

Key words: allenylcyclopentenone, periodate oxidation; Z-carboxymethylenecyclopentenones; intramolecular hydrogen bond.

One of the key steps of our research into the synthesis of structures that could be pharmacologically promising analogs of sea prostanoids of the general formula 1 (X =

CI R CO₂H

NR¹R², OR, SR, Hal, H, etc.; R—alkyl, alkenyl, aryl, etc.) from hexachlorocyclopentadiene^{1,2} is the construction of the exo-carboxymethylene moiety of these compounds. For this purpose, in this study we have synthesized 3-N-substituted 5-allenyldichloro-

cyclopentenones 3a—d from 5-allenyl-2,3,5-trichloro-4,4-dimethoxycyclopentenone (2)³ and studied oxidative cleavage of compounds 3a—d under the action of a RuCl₃·3H₂O—NaIO₄/CCl₄—MeCN—H₂O system.⁴

Although this oxidative system is not highly chemoselective⁵ (RuO₄, which is responsible for the catalyzed oxidative cycle, interacts with alkenes, alkynes, aromatic compounds, alcohols, ethers, etc. in the presence of cooxidants), we demonstrated earlier that the abnormal reaction of the formation of iodohydrin, 2,5-dichloro-5-(2-hydroxy-3-iodopropyl)-4,4-dimethoxy-3-morpholinocyclopent-2-en-1-one, occurs smoothly in the oxidation of 5-allyl-2,5-dichloro-4,4-dimethoxy-3-morpholinocyclopent-2-en-1-one.6 Therefore, the dimethylacetal group, the ring double bond, and other potentially reactive N- and Cl-containing centers in molecules 3a-d should also be tolerant to the action of this oxidizing agent. At the same time, the allene π-system of compounds 3a-d, which is more electrophilic and reactive than the allylic system in 5-allyl-2,5-dichloro-4,4-dimethoxy-3-morpholinocyclopent-2-en-1-one, obviously should more easily undergo attack by RuO4, which is directed preferably at the terminal allene bond for steric reasons. These assumptions were confirmed experimentally. The RuCl3-catalyzed periodate oxidation of cyclopentenones 3a—d proceeded chemo-, regio-, and stereoselectively to give 5Z-carboxymethylenecyclopentenones 4a—d in high yields (Scheme 1).

Scheme 1

CI OME OME OME OME OME
$$Aa-d$$

CI OME $Aa-d$

CI O

The exclusively stereoselective formation of 5Z-isomeric acids $4\mathbf{a}$ — \mathbf{d} engaged our attention: the corresponding 5E-isomers were not found. The Z-orientation of the exocyclic carboxymethylene moiety in molecules $4\mathbf{a}$ — \mathbf{d} was unambiguously confirmed by X-ray diffraction analysis data for one of these compounds, dimethylamino derivative $4\mathbf{d}$. The X-ray diffraction analysis data

indicate that a strong intramolecular H-bond exists between the carboxy group proton and the oxygen of the C(1)=0 moiety, and n- and π -electrons are delocalized throughout the system of conjugated bonds of compound 4d. The chelate H-bond-stabilized 5Z-stereochemistry of the synthesized acids is also confirmed by the absence of COOH group absorption in the region of $3000-3600~\rm cm^{-1}$ in the IR spectra, by the abnormally low-field chemical shifts of the acid proton in the ¹H NMR spectra (δ 15–16) that are independent of dilution, and by the diastereotropism of the protons of NMe₂ (4d) and α , α -CH₂ groups (4a-c).

Undoubtedly the mechanistic aspects of the stereoselective formation of 5Z-acids 4a-d are of interest. Taking into account the facts mentioned above, the possible pathways of oxidative degradation of the allene moiety in compounds 3a-d by a $RuCl_3-NaIO_4$ system (Scheme 2, a and b) should include the formation of the intermediate β -chloroacid 5, which at the moment of its generation is "drawn into" the chelate cycle by the electron-enriched CO group of the chlorovinyl enaminoketone moiety of the molecule. This ensures ease of the reaction and stereospecificity at the step of HCl elimination.

The absence of stereoselectivity in the similar OsO_4 catalyzed periodate oxidation of compound 6 to form a mixture of enals $7,^2$ and isolation of intermediate β -chloroaldehyde 8 in the similar oxidation of the corresponding olefin⁸ (Scheme 3) also confirm the proposed mechanism.

Experimental

IR spectra were recorded on an UR-20 spectrophotometer (as thin layers or as Nujol mulls). ¹H and ¹³C NMR spectra were recorded on a Bruker AM-300 spectrometer (300 and 75.47 MHz, respectively) in CDCl₃ or in (CD₃)₂CO; SiMe₄ was used as the internal standard. Compounds 2, 3a, and 3d were synthesized by earlier reported procedures.^{9,10}

(±)-5-Allenyl-2,5-dichloro-4,4-dimethoxy-3-(N-methylpiperazino)cyclopent-4-en-1-one (3b). A solution of N-methylpiperazine (0.9 g, 9.8 mmol) in MeOH (10 mL) was added to a solution of trichloroallene 2 (1.3 g, 4.6 mmol) (see Ref. 9) in MeOH (30 mL). The reaction mixture was stirred at ~20 °C for 4 to 5 h until the starting compound completely disappeared (TLC). The methanol was evaporated, and water (20 mL) was added to the resulting residue. The products were extracted with CH2Cl2 (4×30 mL). The combined extracts were washed with a saturated aqueous NaCl solution, dried over MgSO₄, and evaporated. Recrystallization of the residue from a pentane-ethyl acetate mixture (1:1) gave compound 3 (1.46 g, 92%), m.p. 78.0-79.5 °C. IR, v/cm⁻¹: 735, 900, 940, 960, 1615, 1650, 1715, 2815. ¹H NMR (CDCl₃), 8: 2.04 (s, 3 H, CH₃N); 3.30, 3.48 (both s, 3 H each, OCH₃); 3.8— 4.0 (m, 8 H, CH₂); 4.39 (dd, 2 H, CH₂, J = 2.6 and 6.6 Hz); 5.42 (t, 1 H, CH=, J = 6.5 Hz). ¹³C NMR (CDCl₃), 8: 45.73 (CH₃N); 48.71 (CH₂N); 51.78, 53.47 (OCH₃); 55.32 (CH_2N) ; 72.62 (C(5)); 80.38 $(=CH_2)$; 92.53 (=CH); 102.83 (C(4)); 105.27 (C(2)); 158.15 (C(3)); 185.37 (C(1)); 208.49 (=C=) Found (%): C, 50.80; H, 5.67; Cl, 20.80; N, 7.90. C₁₅H₂₀Cl₂N₂O₃. Calculated (%): C, 51.87; H, 5.76; Cl, 20.46; N, 8.07

(±)-5-Allenyl-2,5-dichloro-3-diethylamino-4,4-dimethoxy-cyclopent-4-en-1-one (3c) was obtained analogously to compound 3b from trichloroallene 2 (1 g, 3.54 mmol) and diethylamine (0.52 g, 7.07 mmol), m.p. 62.0-63.5 °C. IR, v/cm^{-1} : 745, 910, 1595, 1700, 1960. ¹H NMR (CDCl₃), δ : 1.19 (t, 6 H, CH₃, J = 6.84 Hz); 3.26, 3.39 (both s, 3 H each, OCH₃); 3.72 (m, 4 H, CH₂); 4.95 (dd, 2 H, CH₂=, J = 6.37 and 3.3 Hz); 5.40 (t, 1 H, CH=, J = 6.36 Hz). ¹³C NMR (CDCl₃), δ : 14.05 (CH₃); 51.14, 52.90 (OCH₃); 60.19 (CH₂N); 72.25 (C(5)); 79.85 (CH₂=); 92.67 (CH=); 101.16 (C(4)); 105.0 (C(2)); 157.94 (C(3)); 184.89 (C(1)); 208.09 (=C=). Found (%): C, 50.8; H, 5.8; Cl, 21.96; N, 4.2. C₁₄H₁₉Cl₂NO₃. Calculated (%): C, 52.5; H, 5.93; Cl, 22.19; N, 4.38.

5-(1Z-Carboxymethylene)-2-chloro-4,4-dimethoxy-3morpholinocyclopent-2-en-1-one (4a). RuCl₃·3H₂O (ca. 2 g) was added to a stirred suspension of dichloroketone 3a (see Ref. 10) (0.10 g, 0.3 mmol) and NaIO₄ (0.14 g, 0.65 mmol) in a CCl₄-AcCN-H₂O mixture (2:2:3, v/v). The reaction mixture was stirred for 4-5 h at -20 °C. Then CH₂Cl₂ (10 mL) was added to the resulting mixture, the organic layer was separated, and the water layer was extracted with CH2Cl2 (4×30 mL). The combined organic extracts were dried over MgSO₄, filtered off, evaporated, and the residue was crystallized from diethyl ether. Compound 42 (0.09 g, 95%) was obtained as a light yellow crystalline solid, m.p. 204-206 °C. IR, v/cm⁻¹: 910, 930, 955, 990, 1560, 1715, 2745. ¹H NMR (CDCl₃), 8: 3.29 (s, 6 H, OCH₃); 3.83-3.88 (m, 4 H, CH₂N); 4.08, 4.31 (both m, 2 H each, OCH₂); 6.16 (s, 1 H, =CH). ¹³C NMR (CDCl₃), δ: 49.17, 50.36 (CH₂N); 52.39 (OCH₃); 67.09, 67.34 (CH₂O); 103.93 (C(4)); 109.62 (C(2)); 124.72 (CH=); 137.99 (C(5)); 159.97 (C(3)); 164.82 (CO₂H); 182.24(C(1)). Found (%): C, 49.00; H, 4.85; CI, 11.23; N, 4.50. C₁₃H₁₆CINO₆. Calculated (%): C, 49.21; H, 5.04; Cl, 11.04; N, 4.41.

5-(1Z-Carboxymethylene)-2-chloro-4,4-dimethoxy-3-(N-methylpiperazino)cyclopent-2-en-1-one (4b) was synthesized similarly to compound 4a from ketone 3b (1.46 g, 4.2 mmol) and NaIO₄ (3.90 g, 18 mmol), yield 1.30 g (93%), m.p. 119—121 °C (decomp.) (bright yellow crystals). IR, v/cm^{-1} : 750, 920, 1000, 1605, 1665, 1725, 2750. ¹H NMR ((CD₃)₂CO), δ: 2.75 (s, 3 H, CH₃N); 3.43 (s, 6 H, OCH₃); 3.24—3.50 (m, 4 H, CH₂); 4.26—4.68 (m, 4 H, CH₂); 6.14 (s, 1 H, CH=). ¹³C NMR ((CD₃)₂CO), δ: 42.98 (NCH₃); 45.65, 53.46 (CH₂N); 51.30, 51.82 (OCH₃); 103.61 (C(4)); 109.67 (C(2)); 124.31 (CH=); 137.51 (C(5)); 160.22 (C(2)); 164.00 (CO₂H); 182.34 (C(1)). Found (%): C, 49.75; H, 5.73; Cl, 10.41; N, 8.21. C₁₄H₁₉CiN₂O₅. Calculated (%): C, 50.83; H, 5.75; Cl, 10.74; N, 8.47.

5-(1Z-Carboxymethylene)-2-chloro-3-diethylamino-4,4-dimethoxycyclopent-2-en-1-one (4c) was synthesized similarly to compound 4a from ketone 3c (0.30 g, 0.94 mmol) and NaIO₄ (0.42 g, 1.96 mmol), yield 0.13 g (50%, in view of conversion), m.p. 132—134 °C. IR, v/cm⁻¹: 725, 905, 920, 990, 1640, 1690, 1725, 2770, 3030, 3070. ¹H NMR (CDCl₃), 8: 1.34 (t, 3 H, CH₃, J = 7.01 Hz); 1.39 (t, 3 H, CH₃, J = 7.09 Hz); 3.29 (s, 6 H, OCH₃); 3.97 (q, 2 H, OCH₂, J = 7.05 Hz); 4.0 (q, 2 H, OCH₂, J = 7.08 Hz): 6.19 (s, 1 H, =CH). ¹³C NMR (CDCl₃), 8: 13.7, 15.7 (CH₃); 45.3, 46.92 (CH₂N); 52.17 (OCH₃); 103.93 (C(4)); 109.63 (C(2)); 123.87 (=CH); 138.77 (C(5)); 160.78 (C(3)); 165.16 (CO₂H): 181.46 (C(1)). Found (%): C, 52.20; H, 5.90; Cl. 10.26; N, 4.35.

 $C_{13}H_{18}CINO_5$. Calculated (%): C, 51.49; H, 5.94; Cl, 11.55; N, 4.62.

5-(1Z-Carboxymethylene)-2-chloro-4,4-dimethoxy-3-dimethylaminocyclopent-2-en-1-one (4d) was synthesized similarly to compound 4a from ketone 3d (0.20 g, 0.6 mmol) and NaIO₄ (0.28 g, 1.3 mmol), yield 0.17 g (90%), crystals, m.p. 144–145 °C. IR, v/cm^{-1} : 725, 905, 920, 990, 1640, 1690, 1725, 2770, 3030, 3070. ¹H NMR (CDCl₃), δ : 3.30 (s, 6 H, OCH₃); 3.52, 3.70 (both s, 3 H each, Me₂N); 6.10 (s, 1 H, CH=). ¹³C NMR (CDCl₃), δ : 42.29, 42.45 (Me₂N); 52.04 (OCH₃); 103.53 (C(4)); 110.09 (C(2)); 123.72 (CH=): 138.88 (C(5)); 161.96 (C(3)); 164.99 (CO₂H); 181.74 (C=O). Found (%): C, 46.95; H, 5.04; Cl, 12.01; N, 4.95. C₁₁H₁₄ClNO₅. Calculated (%): C, 47.91; H, 5.08; Cl, 12.89; N, 5.08.

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